



Course Syllabus

| |
|---|
| Department/Faculty School of Engineering |
| Graduate Program Materials Engineering and Nanotechnology |
| Degree <input checked="" type="checkbox"/> Academic Master's <input checked="" type="checkbox"/> Doctorate (PhD) <input type="checkbox"/> Professional Master's |
| Course Name Advanced Electrochemistry |
| Professor(s) Não Preencher |
| Office hours 48 |
| Course Overview <p>The course deals with the basic definitions and concepts of electrochemistry. Models of electrical double layer. Charge transfer kinetics. Butler-Volmer and Tafel empirical equations. Marcus's theory of electron transfer. Activation potentials, ohmic and mass transfer. Faradic electrode processes. Experimental methods for studying solid/liquid interface and the charge transfer kinetics. Electrochemical devices for charge storage, sensors and biosensors with electrochemical detection.</p> |



Topics outline

- 1 – Basic concepts and definitions;
 - 2 - Electrochemical interface: models of electrical double layer formation;
 - 3 – Kinetics of charge transfer; empirical Butler-Volmer and Tafel equations;
 - 4 – Marcus's theory of electron transfer;
 - 5 - Activation, ohmic and mass transfer overpotentials;
 - 6 - Faradic electrode processes;
 - 7 - Electrochemical instrumentation (potentiostatic, galvanostatic and potentiometric techniques)
 - 8- Experimental methods for the study of the solid/liquid interface and charge transfer kinetics (linear and cyclic voltammetry scan rate, pulse voltammetric techniques and electrochemical impedance spectroscopy);
 - 9 – Mechanisms of electrocatalytic processes;
 - 10 - Electrochemical devices for charge storage and chemical sensing (basic principles on the operation of lithium-ion batteries, supercapacitors, sensors and electrochemical biosensors;
- Classes will be expository and experimental.

Letter Grade Assignment

Grade A (Excellent) - Grade points between 9 and 10

Grade B (Good) - Grade points between 8 and 8.9

Grade C (Satisfactory) - Grade points between 7 and 7.9

Grade D (Unsatisfactory) - Grade points between 0 and 6.9



Basic References

COMPTON, R. G; BANKS, C.E. **Understanding Voltammetry**. 3rd Edition. World Scientific Publishing Europe: 2018.

ELIAZ, N; GILEADI, E. **Physical Electrochemistry: Fundamentals, Techniques, and Applications** 2nd ed., John Wiley & Sons, 2019.

BARD, A. J.; FAULKNER, L. R. **Electrochemical Methods: Fundamentals and Applications**. 2nd ed., New York: Wiley, 2001.

BRETT, A. M. O.; BRETT, C. M. A. **Electroquímica: princípios, métodos e aplicações**. Coimbra: Almedina, 1996.

Ciucci, F. Modeling electrochemical impedance spectroscopy, **Current Opinion in Electrochemistry**, v 3, p. 132–139, 2019.

PAJKOSSY, T.; JURCZAKOWSKI, R. Electrochemical impedance spectroscopy in interfacial studies, **Current Opinion in Electrochemistry**, v 1, p. 53-58, 2017.

Supplementary References

MEMMING, R. **Semiconductor Electrochemistry**. 2nd Edition. Germany: John Wiley & Sons, 2015.

Literature Review Articles.